

Determine if the following molecules are Polar or NonPolar by 1/4

VI. Draw the Lewis Dot (Electron) Diagrams for the following Covalent Compounds:

	dot structure	polar/non polar molecule	IMF
1) CH <sub>4</sub>			
2) N <sub>2</sub>			
3) NH <sub>3</sub>			
4) H <sub>2</sub>			
5) H <sub>2</sub> O			
6) Cl <sub>2</sub>			
7) CO <sub>2</sub>			
8) F <sub>2</sub>			
9) CCl <sub>4</sub>			
10) O <sub>2</sub>			